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Of Five Basic Types: Acid-base Reactions, Exchange Reactions, Condensation Reactions (and The Reverse, Cleavage Reactions), And Oxidation-reduction
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Oxidation/reduction (or Redox) Will Be Used In The
Titration Analysis Of An Iron Compound. We Will Use
Potassium Permanganate, KMnO_4 , As The Titrant In
The Analysis Of An Unknown Sample Containing Iron
To Determine The Percent Iron By Mass In The Sample.
In Acidic Solution, Potassium Permanganate Rapidly ...
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Alternative That Best Completes The Statement Or
Answers The Question. 1) The Gain Of Electrons By An
Element Is Called _____. A)oxidation B)sublimation
C)reduction D)disproportionation E)fractionation 2)
_____ Is Reduced In The Following Reaction: $\text{Cr}_2\text{O}_7^{2-} + 6\text{S}_2\text{O}_3^{2-} + 14\text{H}^+ \rightarrow 2\text{Cr}^{3+} + 3\text{S}_4\text{O}_6^{2-} + 7\text{H}_2\text{O}$
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19 REVIEW Oxidation-Reduction Reactions SECTION 1
SHORT ANSWER Answer The Following Questions In
The Space Provided. 1. _____ All The Following
Equations Involve Redox Reactions Except (a) $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2$ (b) $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$ (c) $2\text{HgO} \rightarrow 2\text{Hg} + \text{O}_2$ (d) $\text{SnCl}_4 + 2\text{FeCl}_2 \rightarrow 2\text{FeCl}_3 + \text{SnCl}_2$. 2. Assign
The Correct Oxidation Number To The Individual Atom
Or Ion Below. _____ A. Mn In MnO_2 ... Jan 20th,
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